INTRODUCTION

A significant bottleneck associated with the development of biomarkers for the characterization of disease states involves the analysis of complex molecular mixtures.

1,2 Even after the complexity of the mixture is minimized by purification and separation procedures, it is not uncommon to encounter samples that exceed the peak capacity and dynamic range of the analytical approach.3,4 Additionally, there is increased interest to speed up multidimensional analyses of biological samples and to develop new resources to analyze acquired data.5,6 High-throughput condensed-phase separations techniques such as reversed-phase liquid chromatography and capillary electrophoresis (CE) are often coupled into multidimensional platforms that are used in combination with mass spectrometry to analyze a wide range of proteomic and glycomic samples.5,6

As analyses are extended to include multiple samples from many individuals (or transitioned into a clinical setting where sample turnaround may affect patient outcome), tremendous resources will be required. New methods that could increase the speed of analysis or provide new information about complex molecular systems are likely to be widely adopted.

With this in mind, several laboratories have worked to combine gas-phase ion mobility spectrometry (IMS) with MS for the analysis of mixtures.9,−15 In the nested IMS-MS approach,10 a mixture of ions is produced by electrospray ionization (ESI)16 and is introduced into a drift tube containing an inert buffer gas. As ions drift through the instrument they separate according to differences in their mobilities through the gas.17,18 Because the number density of the gas is much less than that for condensed phases, this separation can be carried out very rapidly (usually in only a few milliseconds). Additionally, the mobility of an ion depends on its shape, such that it is sometimes possible to distinguish different isomers prior to MS analysis (an approach that involves comparisons of experiment with theoretical calculations).19−26

This approach has now been used to investigate many types of biomolecular mixtures, including proteins, peptides, and small molecules.10,15,27−29 Especially topical for the development of biomarkers is the analysis of glycans.30−38 Unlike the linear chains of nucleic or amino acids found in oligonucleotides and peptides, glycans exist as a range of complex branched moieties.39 Thus, in addition to isomers that arise from differences in sequence, variations in branching may also exist. Glycosylation patterns for cancerous and healthy cells are known to exhibit significant differences, including variability associated with the range of expression; the appearance of incomplete (i.e., truncated) structures; accumulation of precursors; and, albeit less frequently, the appearance of novel glycans.38 In the case of N-linked glycans, a number of studies have shown that glycans are larger in tumor cells because of increased branching

ABSTRACT: Altered branching and aberrant expression of N-linked glycans is known to be associated with disease states such as cancer. However, the complexity of determining such variations hinders the development of specific glycomic approaches for assessing disease states. Here, we examine a combination of ion mobility spectrometry (IMS) and mass spectrometry (MS) measurements, with principal component analysis (PCA) for characterizing serum N-linked glycans from 81 individuals: 28 with cirrhosis of the liver, 25 with liver cancer, and 28 apparently healthy. Supervised PCA of combined ion-mobility profiles for several, to as many as 10 different mass-to-charge ratios for glycan ions, improves the delineation of diseased states. This extends an earlier study [J. Proteome Res. 2008, 7, 1109−1117] of isomers associated with a single glycan (S1H3N2) in which PCA analysis of the IMS profiles appeared to differentiate the liver cancer group from the other samples. Although performed on a limited number of test subjects, the combination of IMS-MS for different combinations of ions and multivariate PCA analysis shows promise for characterizing disease states.

KEYWORDS: ion mobility, ESI mass spectrometry, glycans, cancer, principal component analysis
and fucosylation.\textsuperscript{39} In addition, N-linked glycans from cancer cells are also characterized by changes in the linkage, amount, and acetylation of sialic acid.\textsuperscript{39–45}

Mass spectrometry is particularly useful to characterize truncated glycan structures appearing in glycosylation disorders.\textsuperscript{37,46,47} Alterations in the serum glycome associated with ovarian, prostate, pancreatic, breast, and liver cancer have been also analyzed by both ESI-MS and matrix assisted laser desorption ionization MS.\textsuperscript{30,39,48–54} It is becoming clear that MS may play a powerful role in many types of glycomic studies that aim to find serological biomarkers for early diagnosis, prognosis, and treatment of a disease.\textsuperscript{55}

Many experimental factors affect outcome. What would appear to be relatively obvious aspects of analyses become critical when biological (human) samples are involved.\textsuperscript{56} Glycomic studies require sample collection, processing, and handling, as well as glycome purification, MS and statistical analysis (important for interpretation). Hence, there are stringent constraints on protocols required to reach reliable outcomes.

Recently, we combined IMS with MS to analyze serum glycans associated with liver cancer and cirrhosis. In this previous study, we demonstrated that a PCA analysis of a single IMS profile for the \([S_3H_3N_4 + 3Na]^+\) glycan ion was sufficient in some cases to distinguish among diseased states.\textsuperscript{30} Here, we extend these studies by combining and analyzing IMS distributions for numerous glycans with the goal to investigate if such approach could increase our capabilities of distinguishing samples related to different disease states. We find this improves resolution of diseased from healthy individuals. In addition to advantages in delineating physiological states, the use of multiple IMS profiles offers the chance to provide deeper insight into the biochemical origins associated with a disease.

\section*{Materials and Methods}

\subsection*{Materials}

Red top vacutainer blood collection tubes (BD 366430) were obtained from Becton Dickinson (Franklin Lakes, NJ). The Peptide-N-Glycosidase F (PNGaseF; EC3.5.1.52), isolated from \textit{Escherichia coli}, was purchased from Associates of Cape Cod Inc. (East Falmouth, MA). Trifluoroethanol (TFA), 2,5-dihydroxybenzoic acid (DHB), and sodium hydroxide were received from Aldrich (Milwaukee, WI). Chloroform, iodoacetamide, and sodium chloride were from EM Science (Gibbstown, NJ). Dithiothreitol (DTT) and iodoacetamide (IAA) were the products of Bio-Rad Laboratories (Hercules, CA). Ammonium bicarbonate was from Mallinckrodt Chemical Company (Paris, KY), while acetonitrile (ACN) was from Fisher Scientific (Fair Lawn, NJ). All other common chemicals of analytical-grade quality were purchased from Sigma (St. Louis, MO).

\subsection*{Study Population and Sample Collection}

Hepatocellular carcinoma (HC) cases and controls were enrolled in collaboration with the National Cancer Institute of Cairo University, Egypt, from 2000 to 2002, as described previously.\textsuperscript{57} Briefly, adults with newly diagnosed HC, aged 17 and older, without a previous history of cancer, were eligible for the study. Diagnosis of HC was confirmed by pathology, cytology, imaging (CT, ultrasound), and serum alpha-fetoprotein (AFP) levels. Controls were recruited from the orthopedic department of Kasr El Aini Faculty of Medicine, Cairo University.\textsuperscript{58,59} The patients with chronic liver disease, fibrosis and cirrhosis, were recruited from Ain Shams University Specialized Hospital and Tropical Medicine Research Institute, Cairo, Egypt, during the same period.

The diagnosis of liver disease in this group of controls was confirmed by ultrasound-guided liver biopsy. The patients negative for hepatitis B viral infection but positive for HCV RNA and with AFP less than 100 ng/mL were selected for the study. All participants signed informed consent, provided a blood sample, and answered a questionnaire with demographic information, personal habits, medical history, and occupational history. The study protocol was approved by the institutional review committee and conformed to the ethical guidelines of the 1975 Helsinki Declaration.

Blood samples were collected by a trained phlebotomist each day around 10 a.m. and processed within a few hours according to a standardized protocol. Aliquots of sera were frozen immediately after collection and kept at \(-80\) °C until analysis; all IMS measurements were performed on twice-thawed sera.

\section*{Experimental IMS-MS Analysis of Glycans}

\subsection*{Release of N-Glycans from Glycoproteins}

Ninety human serum samples [30 controls (NC), 30 from individuals with hepatocellular carcinoma (HC), and 30 from individuals with cirrhosis \((QT)\) ] were reduced and alkylated as described previously.\textsuperscript{39,53} Briefly, a 10-μL aliquot of serum was added to 150 μL of 25 mM ammonium bicarbonate and 2.5 μL of 200 mM DTT prior to incubation at 60 °C for 45 min. A 10-μL aliquot of 200 mM IAA was added and the reaction was allowed to proceed at room temperature in the dark for 1 h. Subsequently, a 2.5-μL aliquot of DTT was added to react with the excess IAA. The reaction mixture was diluted with 100 μL of ammonium bicarbonate to adjust the pH to 7.5–8.0 for the enzymatic release of N-glycans using PNGase F. Next, a 5 μL aliquot of PNGase F was added to the mixture prior to incubation overnight \((18–22\) h) at \(37\) °C.

\subsection*{Solid-phase Extraction of N-Glycans}

The volume of enzymatically released glycans was adjusted to 1 mL with deionized water and applied to a C18 Sep-Pak cartridge (Waters, Milford, MA), which was preconditioned with ethanol and deionized water as described previously.\textsuperscript{39,53} The reaction mixture was circulated through the cartridge 5-times to retain peptides and O-linked glycopeptides. Glycans were present in the pass-through and the 0.25 mL deionized water washes. The combined eluents were then passed over activated charcoal microcolumns (Harvard Apparatus, Holliston, MA) preconditioned with 1 mL of ACN and 1 mL aqueous solution of 0.1% trifluoroacetic acid (TFA). The microcolumn was washed with 1 mL of 0.1% TFA and samples were eluted with 1-mL 50% aqueous ACN with 0.1% TFA. The purified N-glycans were evaporated to dryness using vacuum CentriVap Concentrator (Labconco Corporation, Kansas City, MO) prior to solid-phase permethylation.

\subsection*{Solid-phase Permethylation}

Permethylation of N-linked glycans stabilizes the structure of sialylated glycans and prevents loss of sialic acid during MS experiments.\textsuperscript{60} It also enables simultaneous analysis of acidic (i.e., sialic acid-containing glycans) and nonacidic glycans in the positive ionization MS mode. Permethylation was performed according to our described procedure.\textsuperscript{60,61} Tubes, nuts and ferrules from Upchurch Scientific (Oak Harbor, WA) were used to assemble the sodium hydroxide capillary reactor. Sodium hydroxide powder was suspended in ACN and packed into Peek tubes (1 mm i.d.; Polymicro Technologies, Phoenix, AZ) using a 100-μL syringe from Hamilton (Reno, NV) and a
syringe pump from KD Scientific, Inc. (Holliston, MA). The sodium hydroxide reactor was conditioned with 60 μL of dimethyl sulfoxide (DMSO) at a 5 μL/min flow rate. Purified N-glycans were resuspended in a 50-μL aliquot of DMSO with 0.3 μL of water and 22 μL methyl iodide. This permethylation procedure has been shown to minimize oxidative degradation and peeling reactions and to eliminate excessive cleanup. The sample was infused through the reactor at 2 μL/min and washed with 230 μL of ACN at 5 μL/min. All eluents were combined while permethylated N-glycans were extracted using 200 μL of chloroform and washed 3 times with 200 μL of water prior to drying.

IMS-MS Analysis and Data Evaluation

Nested-IMS-MS data sets for 90 glycan samples were recorded using a home-built instrument that has been described in detail previously. The specific conditions used in these studies are similar to those reported in our recent study of glycans. General descriptions of theory and experiment that are valuable for understanding the concepts in this paper can be found elsewhere. Briefly, experiments are performed as follows. Glycan ions (usually the +2 and +3 sodiated charge states) are produced by electrospray ionization using a NanoMate (TriVersa, Advion, Ithaca, NY) auto injection system that has been modified to perform the present experiments. The beam of electrosprayed ions is accumulated in an hourglass ion funnel and periodically an electrostatic gate is lowered, releasing a packet of ions into the drift tube that is filled with ~2.5 Torr of 300 K He buffer gas. Under these conditions, the drift tube device affords a resolving power of ~80 to 120. As ions travel through the drift tube they separate based on differences in their mobilities, which are largely defined by the shape and charge state of the ion. Thus, the IMS analysis is sensitive to the isomeric configuration (the covalently bonded configuration) as well as the conformation that the ion adopts upon entering the gas phase. Upon exiting the drift tube, ions are focused into the source region of a time-of-flight mass analyzer and detected using microchannel plates. All of the components, electronics, and computer programs that are used in these experiments have been described previously.

Principal component analysis (PCA) of IMS distributions was done by using MarkerView software (ABI/Scienx, Framingham, MA). Ion mobility profiles were combined by addition of ion mobility profiles to an extended drift time axis. This procedure was done in the same manner for all samples analyzed, that is, ion mobility profiles of the same glycan structures fall across the same range of the drift time index scale. The PCA was done on a total of 81 samples, that is, 25 samples from serum of patients with liver cancer (HC), 28 samples from healthy patients (NC), and 28 samples from patients with the liver cirrhosis (QT). Nine samples were not performed. These 9 samples did appear to electrospray correctly. Supervised PCA was employed. That is, by definition, we used prior knowledge about which sample groups were healthy or diseased. IMS data were weighted using either square root or the natural logarithm of ion intensities across the drift time distributions. Before PCA analysis, intensities were scaled using either autoscaling (each intensity value is subtracted by the average response from all samples and divided by the square root of the standard deviation) or the pareto option (each intensity value is subtracted by the average response from all samples and divided by the square root of the standard deviation).

RESULTS AND DISCUSSION

IMS-MS Distribution of Glycans from Serum

Figure 1 shows an example portion of a two-dimensional (2D) nested-IMS-MS data set (referred to as a t1(m/z) plot) for one of the samples. IMS separation reduces spectral congestion by separating components into mobility families. In the data shown in Figure 1, glycans are found in the families corresponding to doubly- and triply charged ions. Glycan ions are clearly separated from background ions that are commonly observed at m/z values lower than 900. Therefore, integrating the data over narrow regions, where peaks appear, makes it possible to reduce contributions originating from background noise.

Peaks in the spectra are assigned to specific ions (or a range of possible glycan isomers) by comparing the experimental m/z values with theoretical values that are calculated for expected glycans. Over the m/z range between 700 and 1500, the primary
peaks correspond to multiply sodiated glycan ions. Many of the peaks are assigned to the known bi-, tri-, and tetra-antennary glycan structures, with varying degrees of sialylation and fucosylation. Table 1 provides a list of 22 ions (including charge

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“S” represents sialic acid (diamonds); “F” represents fucose (triangles); “H” represents hexose (mannose open circles, galactose solid circles); “N” represents N-acetyl glucosamine (solid squares). "N-linked glycans have been assigned by comparison of measured molecular weights of glycans to those of glycans found in the Consortium for Functional Glycomics database (Functional Glycomics Gateway, available at www.functionalglycomics.org)."

states and assignments) that are observed in this sample as major peaks and correspond to 17 glycan structures. A greater number of glycan ions have been assigned in similar samples using a commercial MALDI-MS instrument. However, the present ESI-IMS-Microwave analysis has been conducted using an acquisition time of 2 min per sample in order to achieve the high throughput that is required to reproducibly analyze a large number of glycan samples. The trade-off in the high throughput analysis is the limited number of glycan ions that can be compared across all data sets. That said, the additional information on gas phase conformations of glycan ions obtained from the drift time distributions of these species is sufficient to distinguish the various phenotypes in the study (see discussion below).

Examples of Glycan IMS Profiles

Figure 2 shows average IMS profiles of glycan ions \([S_1H_2N_4 + 3Na]^{3+}\) \((m/z = 825.9)\) and \([S_1F_1H_2N_4 + 3Na]^{3+}\) \((m/z = 884.9)\). The former ion, \([S_1H_2N_4 + 3Na]^{3+}\), is one of the glycans that is expected (based on our prior work) to be altered as consequence of disease progression, whereas the latter distribution, corresponding to \([S_1F_1H_2N_4 + 3Na]^{3+}\), is a fucosylated glycan which has been implicated in cancer. Interestingly, the IMS distributions for both of these ions show four major features in IMS profiles, in addition to other minor features. The \([S_1H_2N_4 + 3Na]^{3+}\) distribution shows a main peak at 19.50 ms, and three other features at 17.42, 18.32 and 19.17 ms; the \([S_1F_1H_2N_4 + 3Na]^{3+}\) ion shows peaks at 18.66, 19.20, 19.99, and 20.35 ms, respectively. As explained in our previous studies, the observation of multiple peaks in the IMS data could be due to the existence of multiple isomeric forms, and (or) the separation of different conformations (possibly arising from different charge site configurations) of the glycan ions that are stable in the gas phase during the ms time scale of the mobility separation.

It is interesting to plot the data such that it is possible to observe variations in intensities for distributions acquired for individuals. Figure 3 shows this type of representation as dot plots of the intensity observed in IMS for the same structures \([S_1H_2N_4 + 3Na]^{3+}\) \((m/z = 825.9)\) and \([S_1F_1H_2N_4 + 3Na]^{3+}\) \((m/z = 884.9)\) for the three sample types across all individuals. Clearly there is substantial variability in peak intensity (a range of about a factor of 2−5 times) for different individuals within a group.

Another important feature of the present data is that IMS distributions of these and other glycan ions from three groups of samples are highly reproducible. Thorough inspection of glycan IMS profiles shows that there are differences in the relative intensities of specific features. For example, relative intensities of peaks across the IMS profile of \([S_1H_2N_4 + 3Na]^{3+}\) are changing between the sample groups (Figure 2). Since this glycan might exist as a mixture of two positional structural isomers (linkage isomers are not considered) each having multiple conformations, changes in the drift time profiles might be attributed to the differences in isomeric composition of this
glycans are sequentially spliced on an arbitrary drift time index axis. The scheme requires that for all data sets, the extracted drift time distributions of a given glycan begin and end at the same drift time (from a total range of 31.5 ms). These times are assigned identical index numbers such that the same mobilities are compared for all data sets. The separation of different species for each of the 9 m/z values that are chosen is relatively small on this scale. However, as shown in Figure 6, PCA of combined IMS profiles of 10 glycan ions (some of which could exist as multiple isomers) shows distinct separation between sample groups. In contrast, supervised PCA analysis of raw mass spectra of analyzed glycans failed to show any clustering (data not shown). This implies that the ion mobility profiles of glycans can be used to distinguish patient groups in a way that may be more specific than analysis of mass spectra alone. This finding is interesting because it suggests additional evidence that altered expression of glycans is a global cellular process in which multiple glycan structures as well as their conformers and isomers change simultaneously.

To find glycans in which the IMS distributions are most responsible for the separation shown in Figure 6, we examined the corresponding loading plot (data not shown). Drift time points far from the plot origin of the loading plots are responsible for separation of disease groups. Ions having a drift time index near 13S or 156 (i.e., S3F1H6N5 or H5N2, respectively) are primarily responsible for clustering of the ion mobility data from different disease groups. Those two also have the smallest statistical Kruskal–Wallis p values among the 9 glycan structures used as shown in Table 2. All 9 glycan ions have Kruskal–Wallis p values less than 0.05; however, the p values of S3F1H6N5 and H5N2 are 7 × 10⁻⁷ and 1 × 10⁻⁸, respectively.

Evaluation of IMS-MS and PCA Methodology

It is interesting to consider the idea that IMS profiles might be used as specific signatures for qualitative assignment of glycans, perhaps in a manner that is similar to the way that infrared or circular dichroism data are used as signatures for specific structures. In the present data we are interested in differences that are associated with combinations of glycans that may delineate disease state. This combination of approaches suggests that differences are accessible with this type of analysis. Because many variables are involved in these data sets as has been done for other studies. Although differences can be assessed, this approach should be used cautiously as differences may not be informative. It would be interesting to consider the use of these data in the characterization of samples in which the physiological state

![Figure 3. Dot plots of the drift time intensities of [S1H5N4 + 3Na]3+(top) and [S1F1H5N4 + 3Na]3+(bottom) glycans for all samples analyzed in this study. Glycans from serum of individuals with liver cancer (HC, 25 samples) and liver cirrhosis (QT, 28 samples), and glycans originating from serum of healthy individuals (NC, 28 samples) were analyzed. The spacing between the individual samples on the x-axis has been arbitrarily selected in order to display the range in integrated intensities (y-axis) for all samples within a phenotype group. The stars represent minimum and maximum values of drift time intensities and squares represent the average values of drift time intensities in particular sample groups.](image-url)
was unknown. Such experiments would provide a further test of the methodology described here that is under development.

Here we note that recent advances have been made in phenotype differentiation using serum glycan characterization with high-throughput LC-MS techniques. 79,80 Compared to such techniques, the IMS-MS measurement presented here has advantages in throughput. For example a single sample analysis requires less time (5-fold) and does not require additional time for column reconditioning and equilibration as well as sample injection and washing. We therefore present the approach as a very high-throughput means for analyzing large populations of samples. We also note that IMS-MS techniques can readily include nano-LC separations in order to more efficiently separate glycans prior to IMS analysis. 81 Although nano-LC separation would decrease the throughput of the analysis, it could provide an additional degree of separation of glycans and their isomers obtained from human serum. 82 Finally, nano-LC−MS offers excellent sensitivity and selectivity for separation and analysis of glycans, 79,80 while IMS provides information on glycan structures that is not available by MS alone. 28 Therefore, IMS is complementary to high-throughput solution-based separation techniques such as nano-LC and CE, and, in combination with mass spectrometry and statistical analyses, may provide improved delineation of disease states.

Figure 4. (A) Supervised PCA of drift time profiles of [S3F1H6N5 + 3Na]3+. (B) Supervised PCA of combined drift time profiles of [S3F1H6N5 + 3Na]3+ and [S3H6N5 + 3Na]3+. A total of 81 samples are shown (9 samples removed due to nozzle changes).

Figure 5. Combined drift time profiles of the 9 glycans that have demonstrated statistical significance between them (F1H5N4 3+; S1H3N5 3+; S1F1H5N4 3+; S3H6N5 3+; S3F1H6N5 3+; H5N2 2+; and H5N4 3+).
CONCLUSIONS

IMS-MS, combined with supervised PCA is used to analyze glycans isolated from serum of healthy patients, patients with liver cancer, and some with cirrhosis. The IMS data show evidence for multiple reproducible features. In favorable cases, IMS separation of a single m/z ion is sufficient to cluster individuals into cohorts with specific physiologies; however, when IMS profiles for multiple m/z ions are combined the ability to resolve different diseased states based on multivariate PCA of these data improves substantially. Supervised PCA using only the mass spectra generated for these samples from the 2D data sets produced no significant clustering. Thus the IMS-MS (PCA) technology appears to provide a more complete analysis of glycans and to offer significant predictive power with respect to physiological state, at no additional cost in time of analysis. We are presently working on determining a confidence level of the proposed method for the diagnostic analysis of unknown samples.

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